



Optimal configuration for a finite low-temperature source refrigerator cycle with heat transfer law $Q_{\infty}(\Delta T^n)^m$

Jun Li^{1,2,3}, Linggen Chen^{1,2,3}, Yanlin Ge^{1,2,3}, Fengrui Sun^{1,2,3}

¹ Institute of Thermal Science and Power Engineering, Naval University of Engineering, Wuhan 430033.

² Military Key Laboratory for Naval Ship Power Engineering, Naval University of Engineering, Wuhan 430033.

³ College of Power Engineering, Naval University of Engineering, Wuhan 430033.

Abstract

The optimal configuration of a refrigeration cycle operating between a finite low-temperature source and an infinite high-temperature sink are derived by using finite time thermodynamics based on a complex heat transfer law, including Newtonian heat transfer law, linear phenomenological heat transfer law, radiative heat transfer law, Dulong-Petit heat transfer law, generalized convective heat transfer law and generalized radiative heat transfer law, $Q_{\infty}(\Delta T^n)^m$. In the refrigeration cycle model the only irreversibility of finite rate heat transfer is considered. The optimal relation between cooling load and coefficient of performance (COP) of the refrigeration cycle is also derived by using an equivalent temperature of low-temperature source. The obtained results include those with various heat transfer laws and infinite low-temperature source, and can provide some theoretical guidelines for the designs of practical refrigerators.

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Keywords: Finite time thermodynamics; Finite heat capacity reservoir; Refrigeration cycle; Optimal configuration; Optimal performance; Heat transfer law.

1. Introduction

There are two standard problems in finite time thermodynamics [1-20], one is to determine the objective function limits and the relations between objective functions for the given thermodynamic system, and another is to determine the optimal thermodynamic process for the given optimization objectives (i.e. to determine the optimal configurations) for the system which serves as a model for real processes.

It is often the case in practice that the cooling load is generated from heat source which is carried by a finite amount of material with finite heat capacity, rather than from heat extracted from an isothermal, infinite source. In the reversible (infinite-time) limit, the cycle, which extracts the maximum work from a finite heat source, is qualitatively different from the Carnot cycle, and its theoretical efficiency is considerably smaller [21].

The optimal configurations of heat engines under different given conditions were obtained by Ondrechen *et al.* [22], Yan and Chen [23, 24], Xiong *et al.* [25], Chen *et al.* [26-29] and Li *et al.* [30]. Chen [31] discussed the optimal configuration of a class of endoreversible refrigerators, for which only the irreversible heat transfer process is concerned. They derived that the endoreversible Carnot refrigerator is the optimal configuration of these endoreversible refrigerators with Newtonian heat transfer law

according to the maximum coefficient of performance (COP) as the operating goal. Chen *et al.* [32] investigated the effect of heat leakage on the optimal configuration of refrigerator with consideration of finite heat capacity low-temperature source, infinite heat capacity high-temperature sink and Newtonian heat transfer law. Chen *et al.* [33] also given the unified description of endoreversible cycles for linear phenomenological heat transfer law $Q \propto (\Delta T^{-1})$. The results obtained in Refs. [24-27, 30] show that heat transfer law has the significant influences on the optimal configurations and performance of heat engine cycles, and a study on the effect of heat transfer law on optimal configuration of refrigeration cycles is necessary.

This paper will extend the previous work by using a complex heat transfer law, including Newtonian heat transfer law, linear phenomenological heat transfer law, radiative heat transfer law, Dulong-Petit heat transfer law, generalized convective heat transfer law and generalized radiative heat transfer law, $Q \propto (\Delta T^n)^m$, in the heat transfer processes between the refrigerator and its surroundings, to find the optimal configuration of the variable-temperature heat-reservoir refrigeration cycles. In the refrigeration cycle model the only irreversibility of finite rate heat transfer is considered. The optimal relation between cooling load and COP of the refrigerator is also derived by using an equivalent temperature of low-temperature source. The optimal performance of endoreversible and irreversible Carnot refrigerator with infinite thermal-capacity (constant- temperature) heat reservoirs with the same heat transfer law were derived by Li *et al.* [34, 35]. The obtained results of this paper include those with various heat transfer laws and can provide some theoretical guidelines for the designs of practical refrigerators.

2. Refrigeration model

The generalized refrigeration cycle model and its surroundings to be considered in this paper are shown in Figure 1. The following assumptions are made for this model. The system adopted is a working fluid alternately connected to a heat source with finite heat capacity and to a heat sink with infinite heat-capacity. The refrigerator operates in a cyclic fashion with a fixed time τ allotted for each cycle. The low-temperature heat-source is assumed to have a constant heat-capacity C , its temperature is given by $T_x(t)$, and its initial temperature is given by T_L . The high-temperature heat-sink is assumed, for simplicity, to be infinite in size and therefore it has a fixed temperature, T_H . The heat transfer between heat source, heat sink and working fluid obey a complex law, including Newtonian heat transfer law, linear phenomenological heat transfer law, radiative heat transfer law, Dulong-Petit heat transfer law, generalized convective heat transfer law and generalized radiative heat transfer law, $Q \propto (\Delta T^n)^m$. The absorbed and released heats of the working fluid are Q_L and Q_H , respectively.

The two steps in the cycle during which the working fluid is disconnected from one reservoir and connected to another are taken to be reversibly adiabatic. It is assumed that these steps occur instantaneously, which implies that the temperature of the working fluid changes discontinuously.

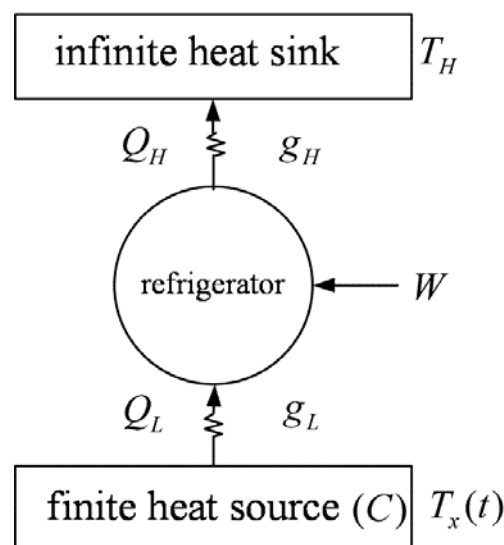


Figure 1. Model of the refrigeration cycle

3. Optimal configuration

Considering that the heat transfer between the refrigerator and its surroundings follows a complex law $Q \propto (\Delta T^n)^m$. Then

$$Q_H = \int_0^\tau g_H(t)[T^n(t) - T_H^n]^m dt \quad (1)$$

$$Q_L = \int_0^\tau g_L(t)[T_x^n(t) - T^n(t)]^m dt \quad (2)$$

where $g_H(t)$ and $g_L(t)$ are heat conductivities between heat sink, heat source and working fluid, respectively. The heat conductivity is product of the overall heat transfer coefficient and corresponding heat transfer surface area of the heat exchanger. It is assume that at $t=0$ the working fluid is in contact with high-temperature heat sink and is separated from the low-temperature heat source by an adiabatic boundary. At a later time $t_1 (0 < t_1 < \tau)$, contact with the heat sink is broken and the working fluid is placed in contact with the heat source. Therefore, one has the flowing relationships

$$g_H(t) = \begin{cases} g_H & 0 \leq t < t_1 \\ 0 & t_1 \leq t < \tau \end{cases} \quad (3)$$

$$g_L(t) = \begin{cases} 0 & 0 \leq t < t_1 \\ g_L & t_1 \leq t < \tau \end{cases} \quad (4)$$

where g_H and g_L are constants.

From the first law of thermodynamics, the work input to the cycle is

$$W = \int_0^\tau \{g_H(t)[T^n(t) - T_H^n]^m - g_L(t)[T_x^n(t) - T^n(t)]^m\} dt \quad (5)$$

From the second law of thermodynamics, the entropy change of the working fluid per cycle is

$$\Delta S = \int_0^\tau \frac{1}{T(t)} \{g_H(t)[T^n(t) - T_H^n]^m - g_L(t)[T_x^n(t) - T^n(t)]^m\} dt = 0 \quad (6)$$

Furthermore, since the heat capacity of the heat source is assumed to be constant, one has

$$dQ_L = -CdT_x(t) \quad (7)$$

Substituting Equation (2) into Equation (7) yields

$$C\dot{T}_x(t) + g_L(t)[T_x^n(t) - T^n(t)]^m = 0 \quad (8)$$

where $\dot{T}_x(t) = dT_x(t)/dt$.

Our problem now is to determine the optimal configuration of the model cycle in which the minimum work input is needed under a given cycle duration τ and cooling load Q_L . Using Equations (5), (6) and (8), one has the modified Lagrangian

$$L = g_H(t)[T^n(t) - T_H^n]^m - g_L(t)[T_x^n(t) - T^n(t)]^m + \frac{\lambda}{T(t)} \{g_H(t)[T^n(t) - T_H^n]^m - g_L(t)[T_x^n(t) - T^n(t)]^m\} + \mu(t) \{C\dot{T}_x(t) + g_L(t)[T_x^n(t) - T^n(t)]^m\} \quad (9)$$

where λ is the Lagrangian constant, and $\mu(t)$ is a function of time. The path for the working fluid which results in the minimum work for a given time interval $\{0, \tau\}$ may now be obtained from the solution of the Euler–Lagrange equations. The Euler–Lagrange equations are given by

$$\frac{\partial L}{\partial T(t)} - \frac{d}{dt} \left[\frac{\partial L}{\partial \dot{T}(t)} \right] = 0 \tag{10}$$

$$\frac{\partial L}{\partial T_x(t)} - \frac{d}{dt} \left[\frac{\partial L}{\partial \dot{T}_x(t)} \right] = 0 \tag{11}$$

For $t_1 \leq t < \tau$, substituting Equations (3), (4) and (9) into Equations (10) and (11), respectively, yields

$$\lambda [T_x^n(t) - T^n(t)] + mnT^n(t) \{ \lambda + [\mu(t) + 1]T(t) \} = 0 \tag{12}$$

$$mng_L T_x^{n-1}(t) [T_x^n(t) - T^n(t)]^{m-1} [T(t) + \mu(t) - \lambda] = CT(t) \dot{\mu}(t) \tag{13}$$

From Equation (12) one can obtain

$$\mu(t) = -\frac{\lambda}{T(t)} - \frac{\lambda [T_x^n(t) - T^n(t)]}{mnT^{n+1}(t)} - 1 \tag{14}$$

The derivative of Equation (14) with respect to t is

$$\dot{\mu}(t) = \frac{\lambda [-nT(t)T_x^n(t)\dot{T}_x(t) + (n+1)\dot{T}(t)T_x^{n+1}(t) + (mn-1)T^n(t)\dot{T}(t)T_x(t)]}{mnT^{n+2}(t)T_x(t)} \tag{15}$$

Substituting Equations (8), (14) and (15) into Equation (13) yields

$$n(m+1)T(t)[T_x^{n-1}(t)\dot{T}_x(t) - T^{n-1}(t)\dot{T}(t)] - (n+1)\dot{T}(t)[T_x^n(t) - T^n(t)] = 0 \tag{16}$$

The solution of Equation (16) is

$$[T_x^n(t) - T^n(t)]T(t)^{-(n+1)/(m+1)} = a(mn) \tag{17}$$

where $a(mn)$ is a constant dependent on mn .

Using the same way of calculation in the case of $t_1 \leq t < \tau$, one can obtain the relation of $T(t)$ and T_H for $0 \leq t < t_1$

$$\lambda T_H^n + \lambda(mn-1)T^n(t) + mnT^{n+1}(t) = 0 \tag{18}$$

Equations (17) and (18) are the major results of this paper. They determine the relation between the temperatures of heat reservoirs and the working fluid. The heat source temperature $T_x(t)$ and the working fluid temperature may be obtained by using Equations (8), (17) and (18), i.e. the optimal configuration of the refrigeration cycles.

4. Effects of heat transfer laws

(1). When $n = 1$, the heat transfer law becomes the generalized convective heat transfer law, Equations (17) and (18) become

$$[T_x(t) - T(t)]T(t)^{-2/(m+1)} = a(m), \quad t_1 \leq t < \tau \quad (19)$$

$$\lambda T_H + \lambda(m-1)T(t) + mT^2(t) = 0, \quad 0 \leq t \leq t_1 \quad (20)$$

where $a(m)$ is a constant dependent on m .

(i). If $m=1$ further, Equations (19) and (20) are the results of the refrigeration cycle with Newtonian heat transfer law. Combining Equations (8), (19) with (20) gives

$$T_x(t) = uT(t) \quad t_1 \leq t < \tau \quad (21)$$

$$T(t) = \begin{cases} vT_H & t_0 \leq t < t_1 \\ T_L \exp[g_L(t-t_1)(u-1)/C] & t_1 \leq t < \tau \end{cases} \quad (22)$$

where u and v are two constants. Equations (21) and (22) are the same results of Refs. [19, 32]. They indicate that the temperatures of heat source and working fluid decrease exponentially with time in the time interval $\{t_1, \tau\}$, and the ratio of the temperatures of the working fluid and heat source is a constant.

(ii). If $m=1.25$, they are the results of the refrigeration cycle with Dulong-Petit heat transfer law [36]. In this case, the heat releasing process is still a constant temperature process. The varying laws of $T_x(t)$ and $T(t)$ in the heat absorbing process become complicate and follow the below relations

$$[T_x(t) - T(t)]T(t)^{-8/9} = a_1 \quad t_1 \leq t < \tau \quad (23)$$

$$C\dot{T}_x(t) = -g_L[T_x(t) - T(t)]^{5/4} \quad 0 \leq t < t_1 \quad (24)$$

where a_1 is a constant.

(2). When $m=1$, the heat transfer law becomes the generalized radiative heat transfer law. Equations (17) and (18) become

$$[T_x^n(t) - T^n(t)]T(t)^{-(n+1)/2} = a(n), \quad t_1 \leq t < \tau \quad (25)$$

$$\lambda T_H^n + \lambda(n-1)T^n(t) + nT^{n+1}(t) = 0, \quad 0 \leq t \leq t_1 \quad (26)$$

where $a(n)$ is a constant dependent on n .

(i). If $n=1$ further, Equations (25) and (26) are the results of the refrigeration cycle with Newtown's heat transfer law, i.e. Equations (21) and (22), they are the same results of Refs. [19, 32].

(ii). If $n=4$, Equations (25) and (26) are the results of the refrigeration cycle with radiative heat transfer law. The temperatures of heat reservoirs and working fluid are complicate and follow the below relations

$$\begin{cases} [T_x^4(t) - T^4(t)]T(t)^{-5/2} = a_2 & t_1 \leq t < \tau \\ C\dot{T}_x(t) = -g_L[T_x^4(t) - T^4(t)] \end{cases} \quad (27)$$

$$\lambda T_H^4 + 3\lambda T^4(t) + 4T^5(t) = 0 \quad 0 \leq t \leq t_1 \quad (28)$$

where a_2 is a constant.

(iii). If $n=-1$, Equations (25) and (26) are the results of the refrigeration cycle with linear phenomenological heat transfer law. Combining Equations (8), (25) and (26) gives

$$T(t) = \begin{cases} \frac{T_L - (ag_L/C)(t - t_1)}{1 - a[T_L - (ag_L/C)(t - t_1)]} & t_1 \leq t < \tau \\ \frac{T_H}{1 - bT_H} & 0 \leq t < t_1 \end{cases} \quad (29)$$

where a and b are two constants.

5. Fundamental optimal relation

Combining the change in the entropy of the working fluid heat absorbing process

$$dS_x = C \ln(1 - Q_L/CT_L) \quad (30)$$

With the condition of internal reversibility, one can introduce an equivalent temperature of the heat source

$$T_L^* = -\frac{Q_L}{dS_x} = -\frac{Q_L}{C \ln(1 - Q_L/CT_L)} \quad (31)$$

And an equivalent temperature of working fluid in the heat absorbing process

$$T_{LC}^* = T_{HC} Q_L / Q_H \quad (32)$$

where T_{HC} is the temperature of working fluid at heat releasing process. Therefore, one can derive

$$Q_L = g_L (T_L^{*n} - T_1^{*n})^m (\tau - t_1) \quad (33)$$

$$Q_H = g_H (T_{HC}^n - T_H^n)^m t_1 \quad (34)$$

$$W = Q_H - Q_L \quad (35)$$

$$\varepsilon = Q_L / W = T_{LC}^* / (T_{HC} - T_{LC}^*) \quad (36)$$

where ε is the COP of the refrigeration cycle.

Defining a ratio of period of two heat exchange processes (f) and the working fluid temperature ratio (x) as follows: $f = t_1 / (\tau - t_1)$, $x = T_{LC}^* / T_{HC}$, where $0 \leq x \leq T_L / T_H$.

Combining Equations (31)-(36) gives the cooling load of the refrigeration cycle as

$$R = Q_L / \tau = \frac{\tau f \varepsilon g_H}{(1 + f)(1 + \varepsilon)} \left\{ \frac{T_L^{*n} - T_H^n [\varepsilon / (1 + \varepsilon)]^n}{[\varepsilon / (1 + \varepsilon)]^n + [fr\varepsilon / (1 + \varepsilon)]^{1/m}} \right\}^m \quad (37)$$

where $r = g_H / g_L$. Taking the derivative of R with respect to f and setting it equal to zero yields

$$f_o = r^{-1/(m+1)} [\varepsilon / (1 + \varepsilon)]^{(nm-1)/(m+1)} \quad (38)$$

Substituting Equation (38) into Equation (37) yields

$$R = \frac{[g_H \varepsilon \tau / (1 + \varepsilon) \{T_L^n (1 + 1/\varepsilon)^n - T_H^n\}]^m}{\{1 + r^{1/(m+1)} [\varepsilon / (1 + \varepsilon)]^{(1-m)/(m+1)}\}^{m+1}} \quad (39)$$

Equation (39) is another major result of this paper. It determines the optimal COP for the fixed cooling load. Since T_L^* in Equations (39) is a function of Q_L , it is independent of Q_L only if C approaches infinite. If C approaches infinite, Equation (39) becomes the fundamental optimal relation between cooling load and COP of an endoreversible Carnot refrigerator coupled to infinite thermal capacity (constant-temperature) heat reservoirs with a complex law $Q \propto (\Delta T^n)^m$ [34].

The relations between the optimal cooling load and COP with different values of m and n are shown in Figure 2. In the numerical calculations, $T_H = 300K$, $T_L = 260K$, $C = 100J/(kg \cdot K)$ and $g_H = g_L = 4W/K^m$ are set. One can see that the heat transfer law has significant influences on the optimal relation between cooling load and COP of the generalized endoreversible refrigeration cycle. The cooling load is a monotonic decreasing function of COP when $n > 0$ and $m > 0$, and a parabolic-like curve when $n < 0$ and $m > 0$.

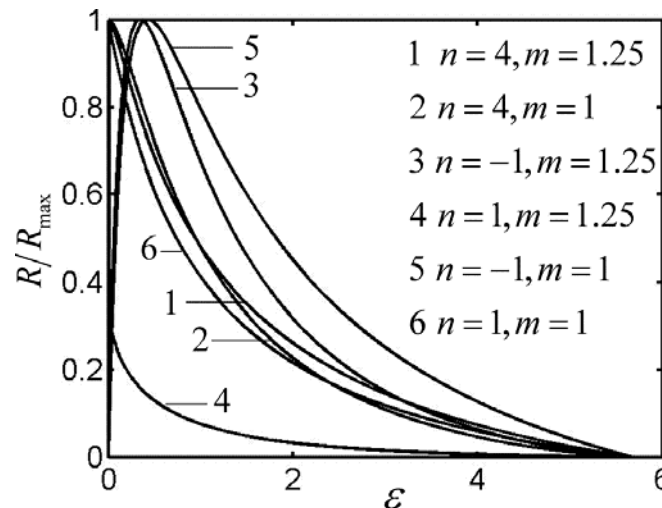


Figure 2. The optimal relation between cooling load and COP of refrigerator with different heat transfer laws

6. Conclusion

The optimal configuration and performance of a refrigeration cycle operating between a finite low-temperature source and an infinite high-temperature sink is studied. In the refrigeration model the only irreversibility of finite rate heat transfer is considered. The heat transfer obeys a complex heat transfer law, including Newtonian heat transfer law, linear phenomenological heat transfer law, radiative heat transfer law, Dulong-Petit heat transfer law, generalized convective heat transfer law and generalized radiative heat transfer law, $Q \propto \Delta(T^n)^m$. The heat transfer law has the significant influence on the optimal relation between cooling load and COP of the generalized endoreversible refrigerator. The cooling load is a monotonic decreasing function of COP when $n > 0$ and $m > 0$, and a parabolic-like curve when $n < 0$ and $m > 0$. The obtained results include those with various heat transfer laws and infinite low-temperature source, and can provide some theoretical guidelines for the designs of practical refrigerators.

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References

- [1] Curzon F L, Ahlborn B. Efficiency of a Carnot engine at maximum power output. Am. J. Phys., 1975, 43(1): 22-24.

- [2] Andresen B, Salamon P, Berry R S. Thermodynamics in finite time: extremals for imperfect heat engines. *J. Chem. Phys.*, 1977, 66(4): 1571-1577.
- [3] Andresen B, Berry R S, Nitzan A, Salamon P. Thermodynamics in finite time I. The step-Carnot cycle. *Phys. Rev. A*, 1977, 15(5): 2086-2093.
- [4] Andresen B, Berry R S, Nitzan A, Salamon P. Thermodynamics in finite time II. Potential for finite -time processes. *Phys. Rev. A*, 1977, 15(5): 2094-2102.
- [5] Gutkowitz-Krusin D, Procaccia I, Ross J. On the efficiency of rate processes. Power and efficiency of heat engines. *J. Chem. Phys.*, 1978, 69(9): 3898-3906.
- [6] Rubin M H. Optimal configuration of a class of irreversible heat engines. I. *Phys. Rev. A*, 1979, 19(3): 1272-1276.
- [7] Rubin M H. Optimal configuration of a class of irreversible heat engines. II. *Phys. Rev. A*, 1979, 19(3): 1277-1289.
- [8] Rubin M H. Optimal configuration of an irreversible heat engines with fixed compression ratio. *Phys. Rev. A*, 1980, 22(4): 1741-1752.
- [9] Salmon P, Nitzan A. Minimum entropy production and the optimization of heat engines. *Phys. Rev. A*, 1980, 21(6): 2115-2129.
- [10] Salmon P, Berry R S. Thermodynamic length and dissipated availability. *Phys. Rev. Lett.*, 1983, 51(13): 1127-1130.
- [11] Andresen B. Finite-Time Thermodynamics. Physics Laboratory II, University of Copenhagen, 1983.
- [12] De Vos A. Efficiency of some heat engines at maximum-power conditions. *Am. J. Phys.*, 1985, 53(6): 570-573.
- [13] De Vos A. Endoreversible Thermodynamics of Solar Energy Conversion. Oxford: Oxford University Press, 1992.
- [14] Bejan A. Entropy generation minimization: The new thermodynamics of finite-size device and finite-time processes. *J. Appl. Phys.*, 1996, 79(3): 1191-1218.
- [15] Berry R S, Kazakov V A, Sieniutycz S, Szwasz Z, Tsirlin A M. Thermodynamic Optimization of Finite Time Processes. Chichester: Wiley, 1999.
- [16] Chen L, Wu C, Sun F. Finite time thermodynamic optimization or entropy generation minimization of energy systems. *J. Non-Equilib. Thermodyn.*, 1999, 24(4): 327-359.
- [17] Chen L, Sun F (eds.). *Advances in Finite-Time Thermodynamics: Analysis and Optimization*. New York: Nova Science Publishers, 2004.
- [18] Chen L. *Finite-Time Thermodynamic Analysis of Irreversible Processes and Cycles*. Beijing: Higher Education Press, 2005. (in Chinese).
- [19] Andresen B. Current trends in finite-time thermodynamics. *Angewandte Chemie International Edition*, 2011, 50(12): 2690-2704.
- [20] Sieniutycz S, Jezowski J. *Energy Optimization in Process Systems and Fuel Cells*. 2013, Oxford, UK: Elsevier.
- [21] Ondrechen M J, Andresen B, Mozurkewich M, Berry R S. Maximum work from a finite reservoir by sequential Carnot cycles. *Am. J. Phys.*, 1981, 49(7): 681-685.
- [22] Ondrechen M J, Rubin M H, Band Y B. The generalized Carnot cycles: a working fluid operating in finite time between heat sources and sinks. *J. Chem. Phys.*, 1983, 78(7): 4721- 4727.
- [23] Yan Z, Chen L. Optimal performance of an endoreversible cycle operating between a heat source and sink of finite capacities. *J. Phys. A: Math. Gen.*, 1997, 30(23): 8119-8127.
- [24] Yan Z, Chen L. Optimal performance of a generalized Carnot cycles for another linear heat transfer law. *J. Chem. Phys.*, 1990, 92(3): 1994-1998.
- [25] Xiong G, Chen J, Yan Z. The effect of heat transfer law on the performance of a generalized Carnot cycle. *J. Xiamen University (Nature Science)*, 1989, 28(5): 489-494(in Chinese).
- [26] Chen L, Zhou S, Sun F, Wu C. Optimal configuration and performance of heat engines with heat leak and finite heat capacity. *Open Sys. Inf. Dyn.*, 2002, 9(1): 85-96.
- [27] Chen L, Zhu X, Sun F, Wu C. Optimal configurations and performance for a generalized Carnot cycle assuming the heat transfer law $Q_{\infty}(\Delta T)^m$. *Appl. Energy*, 2004, 78(3): 305-313.
- [28] Chen L, Sun F, Wu C. Optimal configuration of a two-heat-reservoir heat- engine with heat leak and finite thermal capacity. *Appl. Energy*, 2006, 83(2): 71-81.

- [29] Chen L, Zhu X, Sun F, Wu C. Effect of mixed heat resistance on the optimal configuration and performance of a heat-engine cycle. *Appl. Energy*, 2006, 83(6): 537-544.
- [30] Li J, Chen L, Sun F. Optimal configuration for a finite high-temperature source heat engine cycle with complex heat transfer law. *Sci. China Ser. G: Phys., Mech. Astron.*, 2009, 52(4): 587-592.
- [31] Chen T. optimal configuration of a class of endoreversible refrigerators. *J. Xiamen University (Nature Science)*, 1985, 24(1): 442- 447(in Chinese).
- [32] Chen L, Sun F, Ni N, Wu C. Optimal configuration of a class of two-heat-reservoir refrigeration cycles. *Energy Convers. Manage.*, 1998, 39(8): 767-773.
- [33] Chen L, Bi Y, Wu C. Unified description of endoreversible cycles for another linear heat transfer law. *Int. J. Energy, Environ. Econ.*, 1999, 9(2): 77-93.
- [34] Li J, Chen L, Sun F. Performance optimization for an endoreversible Carnot refrigerator with complex heat transfer law. *J. Energy Institute*, 2008, 81(3): 168-170.
- [35] Li J, Chen L, Sun F. Cooling load and coefficient of performance optimizations for a generalized irreversible Carnot refrigerator with heat transfer law $Q_{\infty}(\Delta T^n)^m$. *Proc. IMechE, Part E: J. Process Mech. Eng.*, 2008, 222(E1): 55-62.
- [36] O'Sullivan C T. Newtonian law of cooling-A critical assessment. *Am. J. Phys.*, 1990, 58(12): 956-960.



Jun Li received all his degrees (BS, 1999; MS, 2004, PhD, 2010) in power engineering and engineering thermophysics from the Naval University of Engineering, P R China. His work covers topics in finite time thermodynamics and technology support for propulsion plants. Dr Li is the author or coauthor of over 30 peer-refereed articles (over 20 in English journals).



Linggen Chen received all his degrees (BS, 1983; MS, 1986, PhD, 1998) in power engineering and engineering thermophysics from the Naval University of Engineering, P R China. His work covers a diversity of topics in engineering thermodynamics, constructal theory, turbomachinery, reliability engineering, and technology support for propulsion plants. He had been the Director of the Department of Nuclear Energy Science and Engineering, the Superintendent of the Postgraduate School, and the President of the College of Naval Architecture and Power. Now, he is the Direct, Institute of Thermal Science and Power Engineering, the Director, Military Key Laboratory for Naval Ship Power Engineering, and the President of the College of Power Engineering, Naval University of Engineering, P R China. Professor Chen is the author or co-author of over 1465 peer-refereed articles (over 655 in English journals) and nine books (two in English).
E-mail address: lgchenna@yahoo.com; linggenchen@hotmail.com, Fax: 0086-27-83638709 Tel: 0086-27-83615046



Yanlin Ge received all his degrees (BS, 2002; MS, 2005, PhD, 2011) in power engineering and engineering thermophysics from the Naval University of Engineering, P R China. His work covers topics in finite time thermodynamics and technology support for propulsion plants. Dr Ge is the author or coauthor of over 90 peer-refereed articles (over 40 in English journals).



Fengrui Sun received his BS Degrees in 1958 in Power Engineering from the Harbing University of Technology, P R China. His work covers a diversity of topics in engineering thermodynamics, constructal theory, reliability engineering, and marine nuclear reactor engineering. He is a Professor in the College of Power Engineering, Naval University of Engineering, P R China. Professor Sun is the author or co-author of over 850 peer-refereed papers (over 440 in English) and two books (one in English).