



Investigation the properties of hip implantation structure based on nanotechnology by using radio frequency magnetron sputtering

Dunya Abdulsahib Hamdi^{1, 2}

1 Department of Prosthetics & Orthotics Engineering Department, Al Nahrain University, Baghdad, Iraq.

2 Surface Analysis and Materials Engineering Research Group, School of Engineering and Information Technology, Murdoch University, Murdoch, WA 6150, Australia.

Received 20 Jan. 2017; Received in revised form 25 Mar. 2017; Accepted 26 Mar. 2017; Available online 1 Nov. 2017

Abstract

In this research, radio frequency magnetron sputtering was used to prepare the Ti6Al4V alloy by powder ceramic Alumina (Al_2O_3 has 1 μm particle size, alpha phase, with purity is 5N (99.999%)) in order to increasing biocompatibility surface alloy. Energy Dispersive X-ray spectroscopy, X-ray diffraction, Scanning Electron Microscopic and Optical microscope were performed to identify phases and microstructure. *In vitro* studies were carried out in simulated body fluid and HP 7.4 with biomimetic tests to see the efficiency of the biocompatibility of the coated surface. Energy Dispersive X-ray spectroscopy chemical analysis showed increasing in Al_2O_3 and a reduction in Ti content with increasing time deposition. X-ray diffraction phase analysis agreement with Energy Dispersive X-ray spectroscopy results, increasing intensity phase (110) and crystalline with heat treatment at 500°C. The Al_2O_3 phase disappeared after immersed and Hydroxyapatite (HAp) phase is very visible with full coated covered the surface with high intensity (002) and (211). Scanning Electron Microscopic showed the Al_2O_3 good coating and there is no crack appearing with convert particle size to 50nm by using radio frequency sputtering, HAp coated all the surface with particle size 300nm after immersion in SBF that is mean the surfaces is biocompatibility and this confirm by Optical microscopy, according to the result the coating will be compatible with human tissue.

Copyright © 2017 International Energy and Environment Foundation - All rights reserved.

Keywords: Ti-6Al-4V alloy; Radio frequency magnetron sputtering; Ceramic coating; Biomaterial; Simulated body fluid and microstructure.

1. Introduction

Biomaterial is defined by the natural or synthetic material that is suitable for interaction into living tissue especially as part of medical device the, they have many characteristics including mechanical, physical, chemical and biological properties that make it suitable for safe, effective, and reliable use within a physiologic environment, as titanium alloy and ceramic represent the type of biomaterial [1]. Titanium and titanium alloys have been extensively studied for many applications in the area of bone tissue engineering due to the light weight, excellent corrosion resistance, high mechanical strength and low elastic modulus. This structure make the titanium alloys used in biomedical devices including screws, hip

and knee prostheses, plates, for either bone replacement or bone fractures [2]. Different methods were done for enhancing the Osseo integration and bioactivity characteristics of the metallic implants such as the coating of the metallic implants with ceramic [3]. Ceramic materials use to cover the implants which used in medical applications, include bioactive ceramics. Al_2O_3 molecule is one of the stable oxides belong to ionic and covalent bonds between Al and O atoms. These strong bonds leave the ceramic unaffected by galvanic reactions (absence of corrosion, e.g. absence of ion release from bulk materials and from wear debris). Wear debris reduces by using femoral heads of Al_2O_3 ceramic bearing against Al_2O_3 cup sockets [4, 5]. For instance, Al_2O_3 has thermal expansion coefficient is very close to that of Ti alloy, the thermal expansion coefficient of Ti alloy substrate is $8.7 \times 10^{-6}/\text{K}$ and for Al_2O_3 is $8.2 \times 10^{-6}/\text{K}$ this lead to reduce mismatch between coated layer and substrate also prevent formation cracks when cooled from the evaluated temperature[6]. Substation many researches has been done for growth alumina thin film with deposition by magnetron sputtering, Since alumina is a very good insulator a radio frequency (RF) alternating current is best method to be used. An advantage of sputtering, is that and the properties of film under control the sputtered flux of atoms will, at steady state conditions, have the same composition as the target, also particle size convert from micro to nanosize during sputtering [7]. The Particle with nanosize main reducing the grainsize of film that is lead to improve in mechanical properties like strength, wear resistance and hardness [8]. The α -alumina (which is the thermodynamically stable polymorph and very good mechanical properties) growth with different methods, typically requires temperatures above 1000°C which limits the choice of substrate material such as titanium alloy to those that can withstand high temperatures. α - Al_2O_3 films was deposit With a negative bias voltage of 200 V and pre-heating the substrate to a lower temperature of 460°C [9]. The parameters of double glow plasma technique was control to prepare α - Al_2O_3 coatings were optimized to get dense and thick α -Alumina coatings on stainless steel 316 L at low temperature of 580°C [10]. RF magnetron sputtering used to study the effect of heat substrate and under layer effect on the structure of Al_2O_3 films [11]. The aim of this study to improve the alumina coating, growth with small nano size and lowering temperature for prosthetics application.

1.1 Materials

The material was used α Al_2O_3 (Aluminum has 1um particle size, alpha phase, with purity is 5N (99.999%)) provided from VTFM(Vacuum Thin Film Materials).The substrate was used titanium alloys (Ti-6Al-4V) GR2 ASTM F136 (Baoji Jinsheng Metal Material Co. Ltd). According to the manufacturer, have a chemical composition show in Table 1.

Table 1. Chemical composition (WT.%).

Ti	Al	V	Fe	C	N	O	H
89.2	5.5-6.5	3.5-4.5	0.40	0.1	0.05	0.20	0.0125

2. Experimental techniques

The Ti-6Al-4V alloy surface modification by deposited biocompatibility of Al_2O_3 , The coatings were deposited on Ti-6Al-4V alloy from Al_2O_3 targets, using an radio frequency(RF) magnetron sputtering system with Ar gas. The as-deposited Al_2O_3 coatings were amorphous, to promote their crystallization, the samples were annealed in an oven for 1 hour at 500°C . The specimens of Ti-6Al-4V alloys were used as substrates in plasma sputtering system with a circular shape of 2 cm diameter and 1.8 mm thickness. The specimen alloys were grained with various grades of SiC paper such as 180, 240, 320, 500, 600, 800, 1000, 1200, 1800 and 2500 μm of grain size and polished using Struers-DAP-U system, Denmark. The polished alloys were etched and ultrasonically cleaned. The vacuum chamber of RF sputtering device was of 1×10^{-7} Torr, and operation frequency generator was 13.65 MHZ with working pressure was 4×10^{-3} Torr with keep all the conditions work constant and change time of sputtering, the process condition was used can be seen in Table2. This condition have been determine by experimental work (pilots study). *In vitro* studies were carried out in simulated body fluid (SBF), biomimetic are based on the growth of calcium phosphate SBF, to produce an apatite layer on the surface of Ti implants, increasing their consequently favouring Osseo integration and osteoconductivity. Passive conducted the biomimetic tests to see the efficiency of the biocompatibility of the coated surface. The biocompatibility experiments were performed by immersing Ti-6Al-4V alloy samples coated with Al_2O_3 single layer for one month in SBF with the compositions shown in Table 3.

Table 2. Deposition conditions of (Al_2O_3) films coated onto Ti-6Al-4V alloy using RF sputtering technique.

Types of deposition single layer	Al_2O_3 layer
Power (W)	200
Working pressure (Torr)	4×10^{-3}
Substrate temperature ($^{\circ}\text{C}$)	250
Distance between target and substrate (cm)	5
time deposition (tD) of Al_2O_3 (hours)	2,4&6

Table 3. The concentration of SBF [7].

Materials	Weight in mg/l
Sodium chloride NaCl	8.036
Calcium chloride CaCl_2	0.293
Sodium bicarbonate NaHCO_3	0.352
Potassium chloride KCl	0.225
Dipotassium hydrogen phosphate trihydrated $\text{K}_2\text{HPO}_4 \cdot 3\text{H}_2\text{O}$	0.230
Magnesium chloride six hydrated $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$	0.311
Sodium sulfate Na_2SO_4	0.072

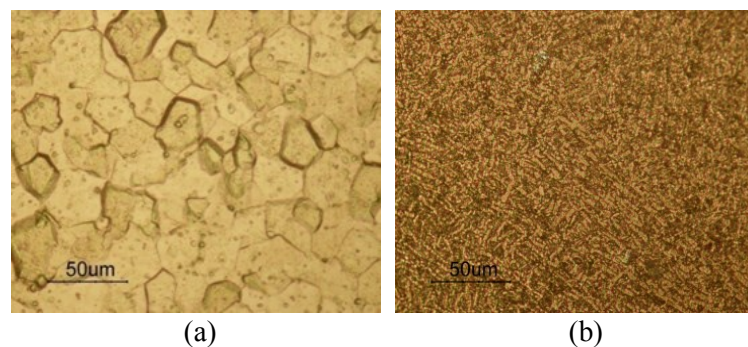
The X-ray Diffraction (XRD- Rigaku Ultima JV, Japan, with 20 KV, 2mA and 0.04KW with a monochromatic radiation scanned $\text{Cu K}\alpha$ (1.54056 \AA)) were performed to measure the identity of the various phases in the films over a 2θ range with a step size of 0.01° . The thickness of coatings deposited onto Ti-6Al-4V alloys was estimated using side view of SEM images (SEM, HITACHI 5-4800 Japan), while the surface morphology of the films were investigated with Scanning Electron Microscopic (SEM). The SEM was operated at 25 kV. The elemental composition of the films was obtained by energy dispersive spectroscopy EDS (EDS, PHILIPS XL series, Japan).

3. Results and discussion

3.1 Before immersion

3.1.1 Microstructure examination (Optical microscopy)

By using the optical microscopy to show that structure of coating in micro size. Figure (1-a) shows the structure of alloy Ti-6Al-4V alloy without coating it is clear from the optical examination that the crystal structure of the alloy is very clear and the grain boundary spirited the grain of Ti alloy. Figure (1-b) shows the structure of the alloy after coating and heat treatment used to obtain the $\alpha\text{-Al}_2\text{O}_3$ phase resulting in relative stability of most common Al_2O_3 particle. The growth of α -alumina phase (Al_2O_3) like finger on the surface of Ti alloy is very clear as shown in the Figure (1-b) after 6 hours and also the morphology of the surface does not show any micro crack and porosity in the coating layer by using RF magnetron sputtering.

Figure 1. The microstructure of Ti6Al4V alloy before and after coated with Al_2O_3 layer.

3.1.2 Energy dispersive X-ray spectroscopy (EDS)

Energy Dispersive X-ray spectroscopy were used to investigate the chemical composition of the Ti-6Al-4V alloy uncoated and coated with Al_2O_3 layers as shown in Figure 2. The energies transition 4.508KeV

and 4.93 KeV belong to Ti $K\alpha$ and Ti $K\beta$ respectively, 0.5 KeV energy belong to O $K\alpha$. The peaks of energy transition of the Al $K\alpha$ that appears at energy of 1.48 KeV which belong to layer of Al_2O_3 coated and substrate. The EDS patterns of the as resived Ti-6Al-4V alloy uncoated shows in Figure 2.a, the intensity: 0.5 for O $K\alpha$ 0.1 for Al $K\alpha$ and 2.58 for Ti $K\alpha$. The chemical analysis report showed low concentration Al_2O_3 is about 3.9 wt% with a high content Ti to 85.5 wt%. Figure 2 (b, c & d) represent Ti-6Al-4V alloy coated with Al_2O_3 at different time for t_D 2, 4 & 6 hours respectively. The increasing in intensity (0.6, 1.5 & 1.7) for Al $K\alpha$ and decreasing in intensity (2, 1.5 & 0.7) for Ti $K\alpha$ belong to increasing the time of deposition. The intensity of O $K\alpha$ increasing (0.4, 0.7 & 0.72) with time deposition belong to oxidation layer. The chemical analysis report in Table 4 showed increasing in Al_2O_3 particles is about 19.7, 27.4 & 30.0 wt% with a reduction in Ti content to 43.5, 29.6 & 25.1 wt%. The increasing of Al_2O_3 particles deposition the reduction of Ti particle [12].

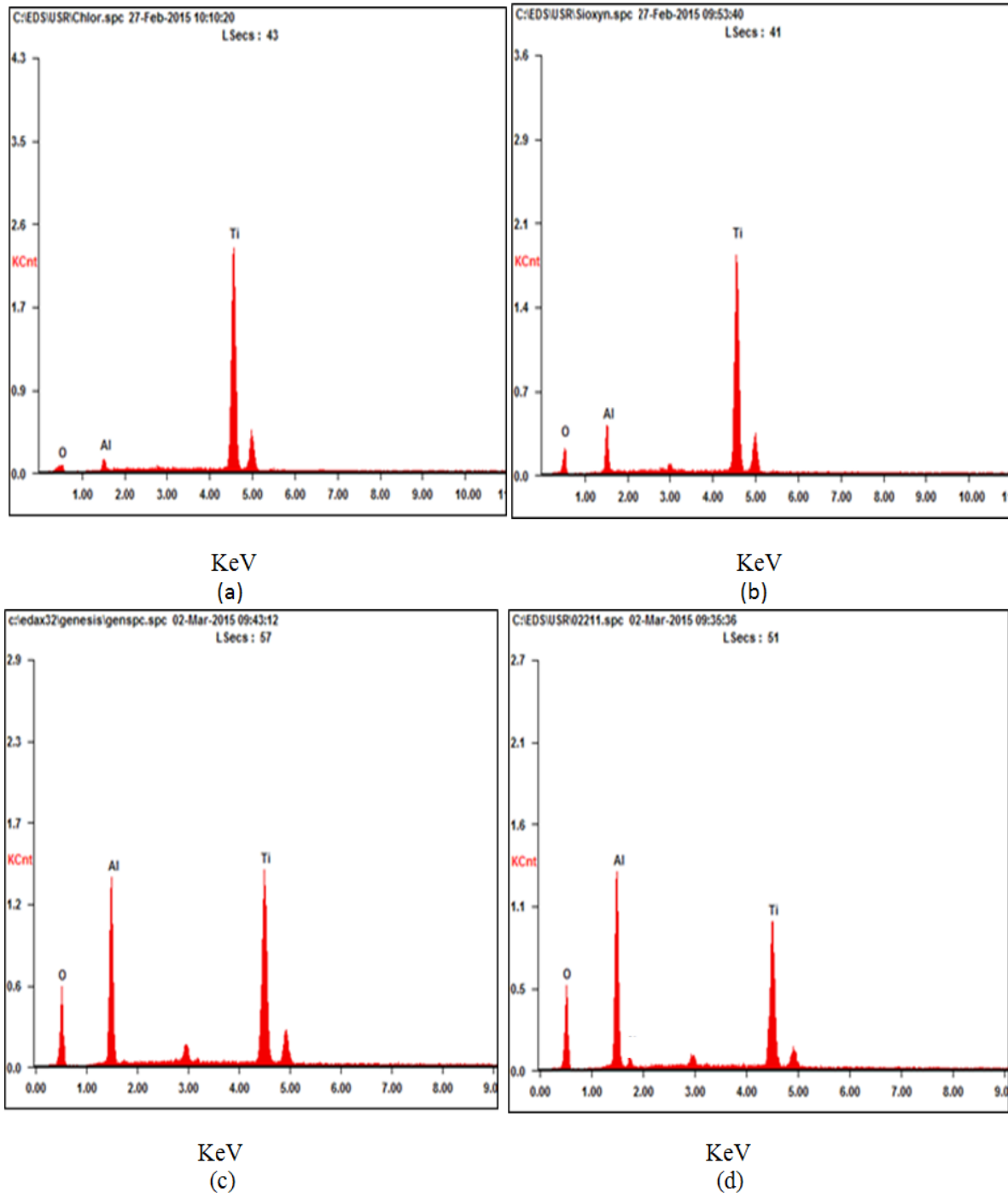


Figure 2. EDS spectra of: (a) Ti-6Al-4V alloy uncoated, (b), (c) & (d) single layer Al_2O_3 for t_D 2, 4 & 6 hours respectively coated on Ti-6Al-4V alloy.

Table 4. Concentrations single layer Al_2O_3 for t_D (2, 4 and 6) hours coated on Ti-6Al-4V alloy determined by EDS.

Exp. No	Hours work	Ti		O		Al	
		Wt%	At%	Wt%	At%	Wt%	At%
uncoated	0	85.54	69.86			3.91	5.67
1	2h	43.57	23.11	36.73	58.34	19.70	18.55
2	4h	29.69	14.58	40.49	59.52	27.48	23.95
3	6h	25.19	11.89	44.23	62.49	30.59	25.62

3.1.3 Scanning electron microscopy (SEM)

The experimental part of other studies was focused on growth of phase α alumina thin films at low-temperature below 1000 °C [7]. Figure 3 shown image of Al_2O_3 coated on Ti-6Al-4V alloy at t_D 6 hours. The morphology of surface was good coating and there is no crack observed because substrate heated during deposition which is one of most important properties of RF sputtering, during deposition process the alumina particle and substrate has high temperature so it solidified slowly. Other authors used plasma spray technique for coating, hot Al_2O_3 particle contact with cold NiAlSi substrate, it solidified very fast which it is lead to crack [18]. The coating growth particles finger-like (like particle which observed by optical Figure (1-b) and diffused is observed after heat treatment, there are no defect in layer but has high porosity. The mechanism via surface diffusion is achieved by the diffusion of atoms along the surface of adjacent particles without shrinkage, this result agreement with Chih-Jen Wang *et al* worked at 1250 °C for centring alumina [8]. The nanosize of alumina with particle size approximately 50nm have strong surface charge this lead to aggregation as shown in Figure 3. The SEM photograph has been choosing for the best result of EDS analysis.

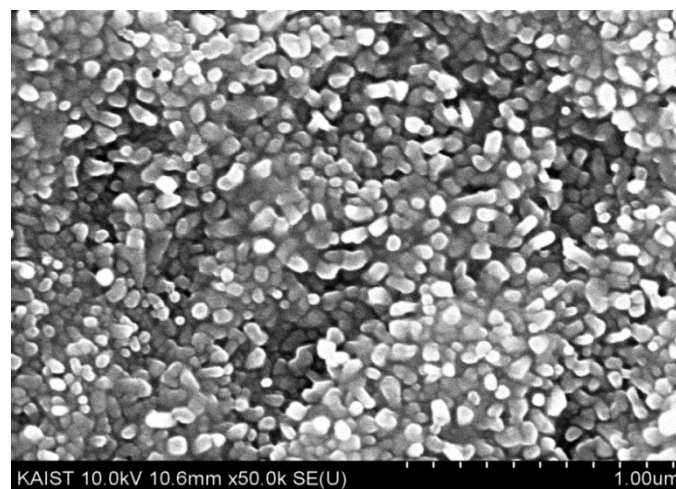


Figure 3. Top-view SEM image of Al_2O_3 coated on Ti-6Al-4V alloy at t_D 6 hours.

3.1.4 X-ray diffraction (XRD)

Figure 4 shows the X-ray Diffraction (XRD) patterns of the precipitation products for investigation of effects of different time deposition Al_2O_3 coated Ti-6Al-4V alloy. The XRD patterns exhibit the crystal major phase presented in alpha alumina ($\alpha\text{Al}_2\text{O}_3$) due to coated layer with diffraction angles (37.7° , 76.8° & 77.2°) belong to planes (110, 1010 & 119) and at increasing time deposition (6 hours) the intensity of plan(110) increasing also appear plan(119). The phase $\delta\text{Al}_2\text{O}_3$ at peak (32.7°) belong to alumina coated [13] was found the thickness of film increasing with increasing the content phase $\delta\text{Al}_2\text{O}_3$. The different phase alumina appearing at increasing the thickness of film [14]. This results agreement with X-ray Diffraction XRD pattern of plasma sprayed AlN powder into distilled water [15]. Titania (TiO_2) at angles (32°) with planes (101) belong to natural oxide layer of alloy consist of (TiO_2 , Ti_2O_3 and TiO) [16]. It was observed that by increasing in time deposition, the intensity of signals phase Al_2O_3 are increased and agreement with EDS. In this experimental with RF sputtering the peak energy $\alpha\text{Al}_2\text{O}_3$ is higher the

same phase result from Al_2O_3 -Al composite coatings deposited by plasma spraying [17]. The sharp peaks belong to high crystal and nano size particle alumina and this agreement with result of SEM.

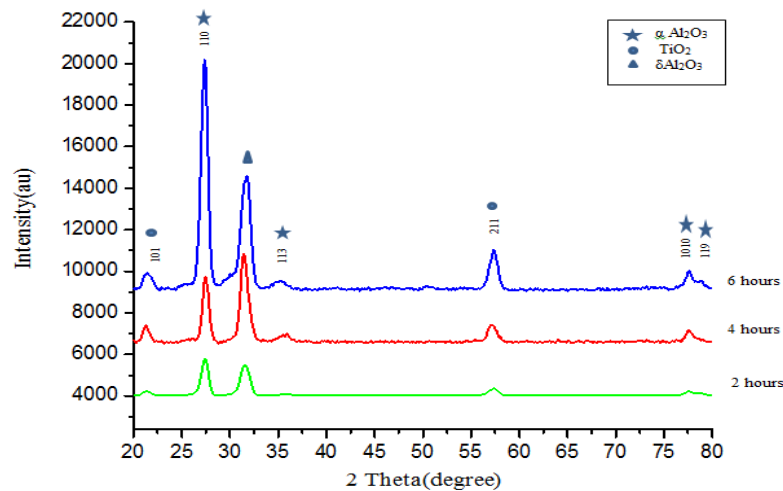


Figure 4. XRD patterns of the Al_2O_3 coated Ti6Al4V alloy for 2,4&6 hours deposition time t_D .

3.2 After immersion in SBF

3.2.1 Microstructure examination (Optical microscopy)

Figure 5 shows the microstructure immersing Ti-6Al-4V alloy samples coated with Al_2O_3 single layer for one month in SBF and the layer of HAp is very visible with full coated, the substrate and Al_2O_3 layer approximately disappear. The composite of HAp in solution of SBF attractive to the surface of sample and great thick layer from HAp as shown in Figure 5 this mean the surface is biocompatibility.

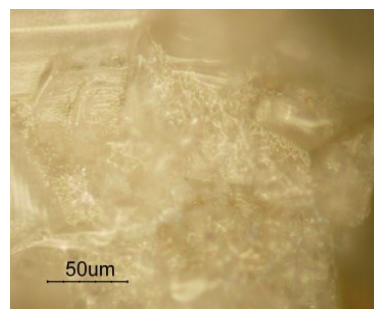


Figure 5. The microstructure of Ti-6Al-4V alloy coated with Al_2O_3 layer after immersion in SBF, with full coated HAp layer.

3.2.2 Scanning Electron microscopy (SEM)

The HAp growth on surface of sample this means the surface with biomaterial properties. Apatite was present as aggregates, showed different shapes as rough, granular to dense and its particles, thick-like plates as shown in Figure 6 like shape particles precipitation was observed [4, 20]. Also the particle with nano size approximately 300nm and uniform grain size with a narrow size distribution corresponding to the crystallinity improvement of the HAp powders after heat treatment at 500°C.

3.2.3 X-ray diffraction (XRD)

From Figure 7 the Al_2O_3 peaks disappear due to the high thick of Hydroxyapatite (HAp) layer and the strong HAp was at (112) and (002) at (32.19 and 25.86) respectively with high intensity which demonstrate that the most stable phase is HAp, this mean an improvement in the coating crystallization Al_2O_3 and has good bonds between the particles of Al_2O_3 and HAp by using plasma sputtering. The Al_2O_3 (110) at Figure 4 dissolved and disappear after immersed the sample in SBF. Also other peaks for HAp at (102), (300) and (421) belong to $\text{Ca}_4\text{H}(\text{PO})_3 \cdot 2.5\text{H}_2\text{O}$ other calcium phosphor compound phases. This preferred orientation suggested that high coating and chemical reaction between the different

components were occurred. The purpose from immersing the samples in SBF is to attract the apatite species (Ca &P) spectra from solution. This result is in a good agreement finding [19].

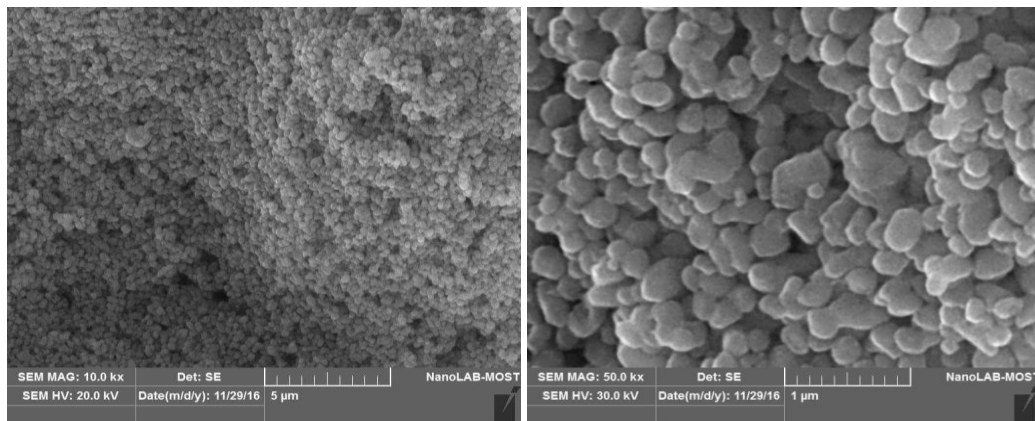


Figure 6. Top-view SEM image of Al_2O_3 coated on Ti-6Al-4V alloy at t_D 6 hours after immersion.

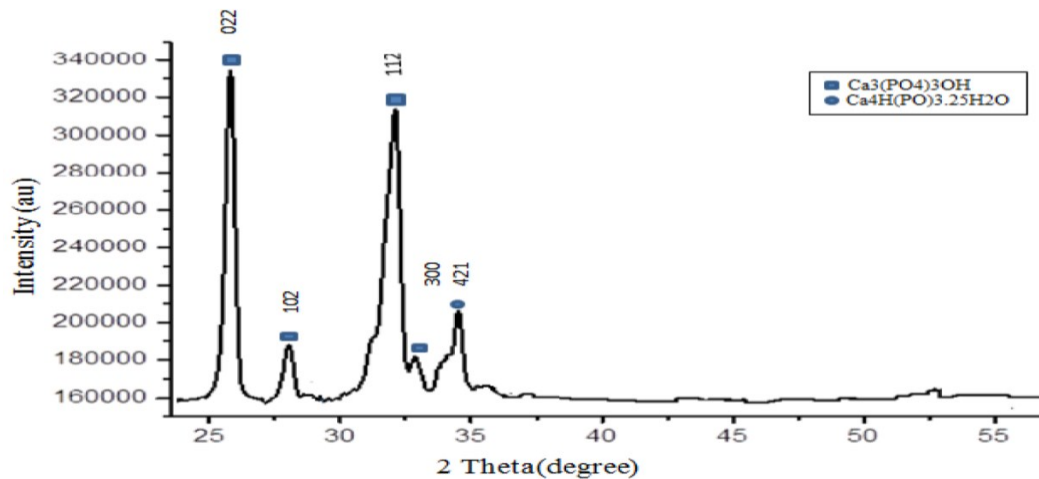


Figure 7. XRD patterns of the Al_2O_3 coated Ti-6Al-4V alloy for 2, 4 & 6 hours deposition time t_D after immersion.

4. Conclusions

1. X-ray diffraction shows alumina Al_2O_3 and base metal (Ti-6Al-4V) peaks disappear due to high thickness of coating.
2. There is no crack observe by SEM after deposition because the substrate heated before and during coating process.
3. The alumina nano size particles have strong surface charge which lead to aggregation. The SEM shows different shape of aggregate it is particles like thick plates, the aggregate depend on size of particle and coating method.
4. Uniform grain size distribution due to crystalline improvement after heat treatment at 500 °C.
5. The surface biocompatibility improvement and the piratical size of Al_2O_3 converted from micro to nano size, this agrees with nano size of cell human. According to the result the coating will be compatible with human tissue.

References

- [1] Russell Wanhill, Simon Barter ' Fatigue of beta processed and beta heat-treated titanium alloys' Springer, New York, USA, 2012.
- [2] Kun Mediaswanti, Cuie Wen, Elena P. Ivanova, Christopher C. Berndt and James Wang ' Sputtered Hydroxyapatite Nano coatings on Novel Titanium Alloys for Biomedical Applications' Intch,2013.

- [3] A. Vladescu, M. A. Surmeneva, C. M. Cotrut, R. A. Surmenev, I. V. Antoniac' Bioceramic Coatings for Metallic Implants' Bioceramics and Biocomposites Sprinker, 15 October 2015.
- [4] Aldo R Boccaccini, P.X. Ma 'Tissue Engineering Using Ceramics and Polymers' Wood head Publishing series in Biomaterial, 2nd edition Elsevire, 2014.
- [5] Rosario Pignatello 'Biomaterials Applications for Nanomedicine' Material since, 2011.
- [6] Zhou-Cheng Wang, Yong-Jin Ni and Jin-Cong Huang' Fabrication and characterization of HAp /Al₂O₃ composite coating on titanium substrate' Journal of Biomedical Science and Engineering, Vol.1, 2008.
- [7] Jon Martin Andersson 'Controlling the Formation and Stability of Alumina' Linqabaings University institute of technology, Sweden, 2005.
- [8] Chih-jen Wang, Chy-Yuen Huang, Yu-Chon.Chun Wu ' Two-step sintering of fine alumina–zirconia ceramics' Ceramics International, Elsever, 2008.
- [9] E Wallin, E P Munger, V Chirita, U Helmersson 'Low-temperature α -alumina thin film growth: ab initio studies of Al adatom surface migration' Journal of Physics D: Applied Physics, Vol.42, No.12, 2009.
- [10] I. Mohammad sadiq, Basodev Chowdhury, BTech 'Antimicrobial sensitivity Eschrichia coli to alumina nano particles ' Nanomedecin Elsever Vol.5, 2009.
- [11] T. Kohara, H. Tamagaki, Y. Ikari, H. Fujii' Deposition of α -Al₂O₃ hard coatings by reactive magnetron sputtering' Surface and Coatings Technology, Vol. 185, Issues 2-3, 2004.
- [12] S. Karthikeyan and L. Vijayaraghavan' Influence of Nano Al₂O₃ Particles on the Adhesion, Hardness and Wear Resistance of Electroless NiP Coatings' International Journal of Materials, Mechanics and Manufacturing, Vol.4, No.2, 2016.
- [13] T.V. Thamaraislvi and S. Rajeswari 'Biological Evaluation of Bio ceramic Materials A-Review ' Trends Biomater. artf.organs, Vol.18, No.1, 2004.
- [14] R. Tomaszek, L. Pawlowski, J. Zdanowski, J. Grimblot Grimblot 'Microstructural transformation of TiO₂, Al₂O₃+13TiO₂ and Al₂O₃+40TiO₂at plasma spraying and laser engraving'. Surface coating technology, Vol.185, No.2, 2004.
- [15] H. Yang, W. Luan and S. Tu 'Corrosion Behavior and Thermal Conductivity of Plasma Sprayed AlN/Al₂O₃ Coating' Materials Transactions, Vol.47, 2006.
- [16] T. Zubaydi, U. Kamachi, S. Rajagopalan, R. Asokamani, Baldev Raj, Surface Characterization of passive film formed on nitrogen ion implanted Ti-6Al-4V and Ti-6Al-7Nb alloys using SIMS, Journal of Corrosion Science 45, 2003, pp. 1951-1967.
- [17] Z. Yin, S. Tao, X. Zhou, C. Ding 'Microstructure and mechanical properties of Al₂O₃–Al composite coatings deposited by plasma spraying' Applied Surface Science Vol.254, 2008.
- [18] P. Avram, M. Imbrea, B. Istrate and C. Munteanu 'Properties of Al₂O₃and NiAlSi coatings obtained by atmospheric plasma sprating on 34CrNiMo6 substrate' Indian Journal of Engineering and Materials Sciences, Vol.21, 2014.
- [19] Y. Liu, B. Huang., J. Ruand and Y. HE 'Behaviour of composition Ca/P bioceramics in simulated body fluid' Journal of Mater Sci. Technol. Vol.14, 1994.
- [20] M. Santos, Oliveira, L. Souza, H. Mansur, W. Vasconcelos 'Synthesis Control and Characterization of Hydroxyapatite Prepared by Wet Precipitation Process' Material Research, Vol.4, 2004.